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13.2 The Nature of Liquids The general properties of solids reflect the orderly arrangement of their particles and the fixed locations of their particles. In most solids, the atoms, ions, or molecules are packed tightly together. When you heat a solid, its particles vibrate more rapidly as their kinetic energy increases.

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Activities for chapter 13: States of matter •What do I already know about states of matter?(index card) and Vocabulary table •Chapter 13 reading guide (feb break assignment) and Powerpoints •POGIL activities: (1) Kinetic Molecular Theory; (2) Phase Changes; (3) Intermolecular forces of attraction (Group Work in class week of feb 14 and finish for HW)

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Chapter Topics. You'll learn all of the chemistry topics covered in the textbook chapter, including: The kinetic molecular theory Pressure The Boltzmann distribution

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